

Metals that form more than one cation.

Cr Fe Co
+2/+3

Cu Hg
+1/+2

Sn Pb Mn
+2/+4

	Element	Ion	Classical Name	Stock Name
Cr Fe Co +2/+3	Chromium	Cr^{2+}	Chromous ion	Chromium (II) ion
		Cr^{3+}	Chromic ion	Chromium (III) ion
	Iron	Fe^{2+}	Ferrous ion	Iron (II) ion
		Fe^{3+}	Ferric ion	Iron (III) ion
Cu Hg +1/+2	Cobalt	Co^{2+}	Cobaltous ion	Cobalt (II) ion
		Co^{3+}	Cobaltic ion	Cobalt (III) ion
	Copper	Cu^+	Cuprous ion	Copper (I) ion
		Cu^{2+}	Cupric ion	Copper (II) ion
Sn Pb Mn +2/+4	Mercury	Hg_2^{2+}	Mercurous ion	Mercury (I) ion
		Hg^{2+}	Mercuric ion	Mercury (II) ion
	Tin	Sn^{2+}	Stannous ion	Tin (II) ion
		Sn^{4+}	Stannic ion	Tin (IV) ion
Sn Pb Mn +2/+4	Lead	Pb^{2+}	Plumbous ion	Lead (II) ion
		Pb^{4+}	Plumbic ion	Lead (IV) ion
	Manganese	Mn^{2+}	Manganous ion	Manganese (II) ion
		Mn^{4+}	Manganic ion	Manganese (IV) ion