Naming Acids

- ▶ To name acids of oxy-ions, replace
 - the "-ate ion" with "-ic acid"
 - the "-ite ion" with "-ous acid"

Р	PO ₄ 3-	Phosphate ion	H ₃ PO ₄	Phosphoric acid
	PO ₃ 3-	Phosph ite ion	H ₃ PO ₃	Phosphorous acid
S	SO ₄ ²⁻	Sulf ate ion	H ₂ SO ₄	Sulfuric acid
	SO ₃ ²⁻	Sulf ite ion	H ₂ SO ₃	Sulfur <mark>ous acid</mark>
С	CO ₃ ²⁻	Carbonate Ion	H ₂ CO ₃	Carbon <mark>ic acid</mark>
	CO ₂ ²⁻	Carbonite Ion	H ₂ CO ₂	Carbonous acid
N	NO ₃ 1-	Nitr <mark>ate</mark> Ion	HNO ₃	Nitric acid
	NO ₂ 1-	Nitr <mark>ite</mark> Ion	HNO ₂	Nitrous acid
Cl,Br, I	ClO ₄ 1-	<i>Per</i> chlorate Ion	HClO ₄	Perchloric acid
	ClO ₃ 1-	Chlorate Ion	HClO ₃	Chloric acid
	ClO ₂ 1-	Chlor <mark>ite</mark> Ion	HClO ₂	Chlorous acid
	ClO ¹⁻	<i>Hypo</i> chlor ite Ion	HClO	Hypochlorous acid